

# **Chapter 5 Review The Periodic Law Section 1**

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Section 1

Chapter 5 Review The Periodic Law

CHAPTER 5 REVIEW The Periodic Law

SECTION 3 SHORT ANSWER Answer the following questions in the space provided.

1. c When an electron is added to a neutral atom, energy is (a) always absorbed. (c) either absorbed or released. (b) always released. (d) neither absorbed nor released.

2. d The energy required to remove an electron from a neutral atom is the atom's ionization energy.

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electrons that can fill the outer sublevels of the elements of that period. 6. An element's period corresponds to its highest occupied main energy level. 7. a. the type of sublevel being filled in Math Tutor Writing Electron Configurations All of seat and table review from Elcho. Chapter 5 Section 2 The Modern Periodic Table Answers.

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2. Chapter 5 Review The Periodic Law Answers 5 1 Modern Chemistry 37 The Periodic Law CHAPTER 5 REVIEW The Periodic Law SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. \_\_\_\_\_ When an electron is added to a neutral atom, energy is (a) always absorbed. (c) either absorbed or released. (b) always



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released. (d) neither absorbed nor released. Modern Chemistry 37 The

Periodic Law CHAPTER 5 REVIEW

The ... 1. a. The group 14 element in Period 4 - Ge [AR]  $4s^2 3d^{10} 4p^2$

b. The only metal in Group 15 - Bi [Xe]  $6s^2 4f^{14} 5d^{10} 6p^3$  c. The

transition metal with the smallest atomic mass - Sc [Ar]  $4s^2 3d^1$

d. What are the answers to Chapter 5 Review Periodic Law 5-2 ... Title:

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